

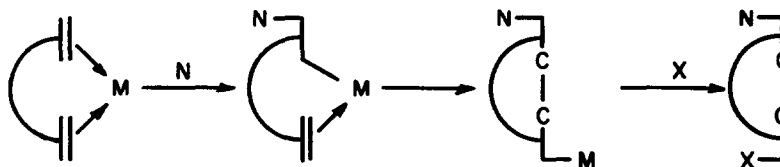
CONVERSION OF CYCLOOCTA-1,5-DIENE-PALLADIUM (II) CHLORIDE TO 6-EXO-CHLORO-2-ENDO-ACETOXYBICYCLO [3 3 0] OCTANE BY LEAD TETRAACETATE IN ACETIC ACID

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In connection with a general synthetic scheme where dienes are converted to carbocyclic compounds via transition metal-diene complexation as illustrated below, we examined an oxidative decomposition of 1,5-cyclooctadiene-PdCl₂ complex. It was recently reported that oxidation of 1,5-cyclooctadiene by PdCl₂-Pb(OAc)₄ in HOAc gave 2,6-diacetoxy bicyclo [3 3 0] octane (4), exclusively as the di-endo isomer.¹ The authors rationalized that the π-complex 1 formed under the reaction conditions, underwent trans-nucleophilic addition of acetate followed by cis-insertion of Pd(II) and another S_N2 substitution by acetate

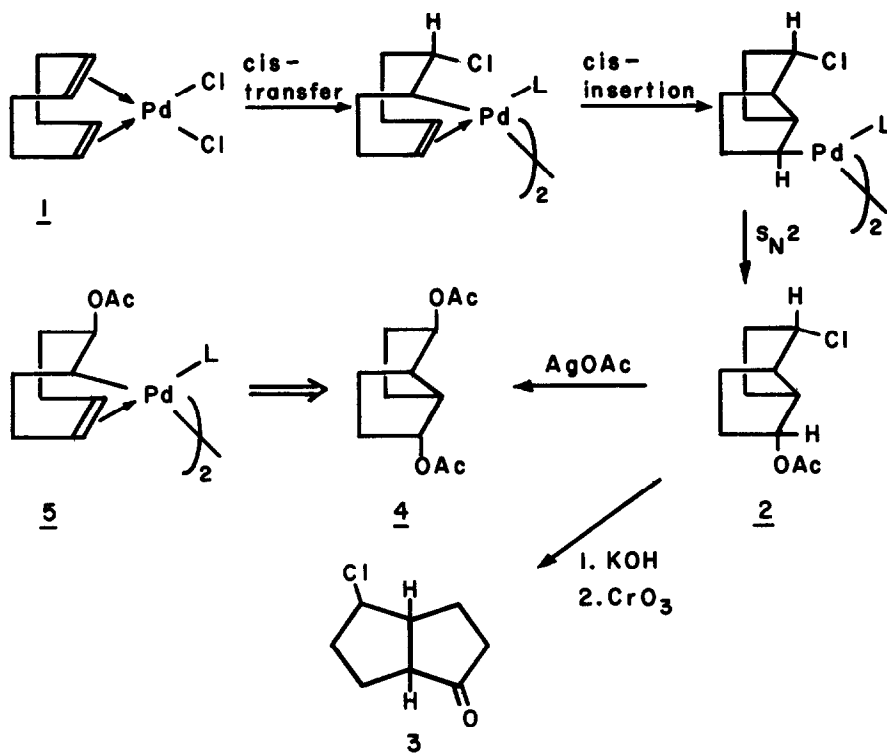


When the preformed 1,5-cyclooctadiene-PdCl₂ complex (1)² was treated with Pb(OAc)₄ in HOAc at r t for 20 hr, the major product isolated in 60% yield was not 4 but 6-exo-chloro-2-endo-acetoxy bicyclo [3 3 0] octane (2) [ir, 1725 cm⁻¹, nmr δ 1.2-2.1 (m, 8H), 2.14 (s, 3H), 2.9 (br m, 2H), 4.2 (q, 1H), 5.4 (q, 1H), m/e 202, 187, 167, 159, 142] The chloroacetate 2 was hydrolyzed (KOH/MeOH, r t) and oxidized (CrO₃/H₂SO₄ - acetone) to give the chloro ketone 3 [ir, 1735 cm⁻¹, nmr δ 1.6-2.6 (m, 8H), 3.0 (m, 2H), 4.2 (m, 1H), Analysis found C, 60.49, H, 6.89, Cl, 22.29 C₈H₁₁O Cl requires C, 60.57, H, 6.94, Cl 22.02%]

The nmr spectrum of 2 showed considerably smaller couplings for the carbonyl proton at δ 4.2 than the carbonyl H at δ 5.4 suggesting that the Cl substituent has the exo stereochemistry. Definitive evidence for the structure 2 was obtained by its smooth conversion to the known 4 with AgOAc in refluxing HOAc.

When 1,5-cyclooctadiene was treated with PdCl₂-Pb(OAc)₄ as reported¹, the presence of 2 in ca 10% yield could be detected. These results seem to suggest that σ-bonded complex 5 rather than π-complex 1 was formed as the major intermediate in the PdCl₂-Pb(OAc)₄ conditions, and that cis-intramolecular ligand transfer from the metal to the olefin carbon is considerably faster than the well-established addition of nucleophile to olefin-Pd(II) complex^{3,4} under these conditions.

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